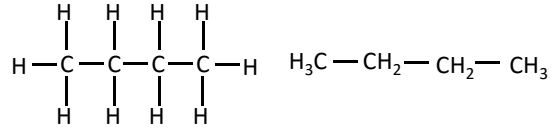


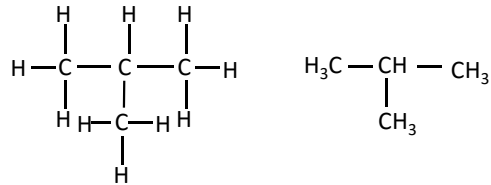
Ex 119C.

a.

${}_1H(K)^1$ manque 1 e⁻ (règle du duet) forme une liaison : H—
 ${}_6C(K)^2 (L)^4$ manque 4 e⁻ (règle de l'octet) forme 4 liaisons : C \equiv

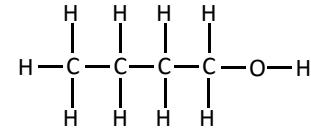


b.

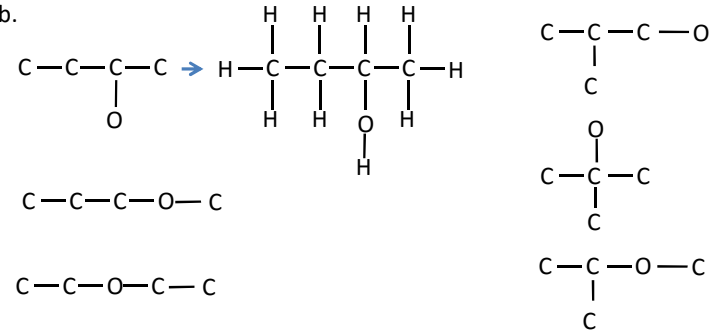
**Ex 120C.**

a.

${}_8O(K)^2 (L)^6$ manque 2 e⁻ (règle de l'octet) forme 2 liaisons : O \equiv



b.

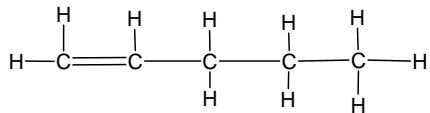
**Ex 121C.**

1.

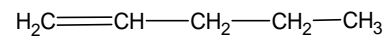
H: 1 liaison

C: 4 liaisons

2.



3.



4.

