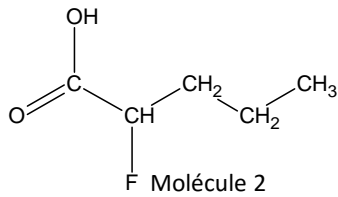
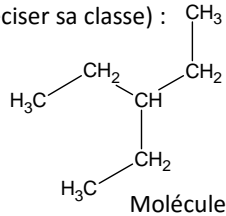


Exercice 1 : Faire la représentation de Lewis (ou formule développée) des molécules suivantes (s'il s'agit d'un alcool préciser sa classe) puis préciser le groupe (l'écrire et l'entourer) et la fonction de la molécule

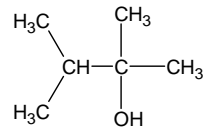
Donnée: ${}_{7}^{14}\text{N}$

- 2-bromo-3-méthylpentanal.
- acide 2-méthylpronoïque
- 1,1-dichlorobutan-2-one
- 3-iodohexane-2-ol
- 1-fluoropentanol
- NH_2Cl

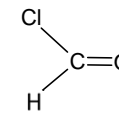
Exercice 2 : Écrire le nom des molécules suivantes (s'il s'agit d'un alcool préciser sa classe) :



Molécule 3

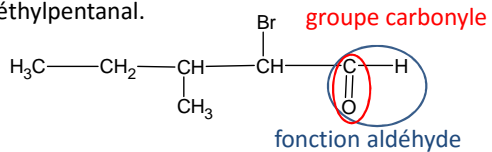


Molécule 4

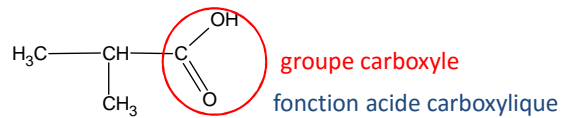


Exercice 1 :

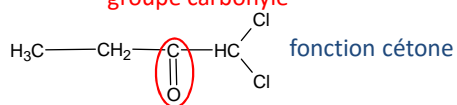
2-bromo-3-méthylpentanal.



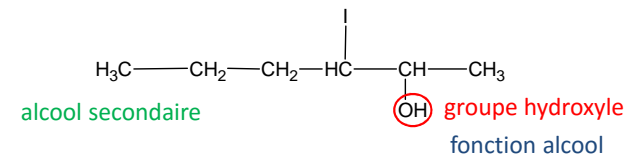
2. acide 2-méthylpronoïque



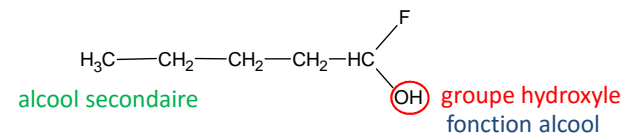
3. 1,1-dichlorobutan-2-one



4. 3-iodohexane-2-ol

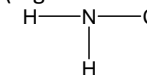


5. 1-fluoropentanol



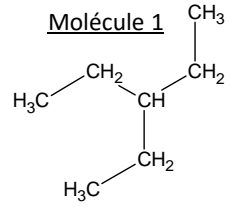
6. NH_2Cl

${}_{7}^{14}\text{N}(\text{K})^2(\text{L})^5$: besoin de 3 e- (règle de l'octet) donc N forme 3 liaisons.



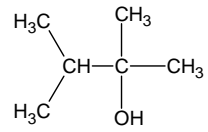
Exercice 2:

Molécule 1



2-éthylpentane

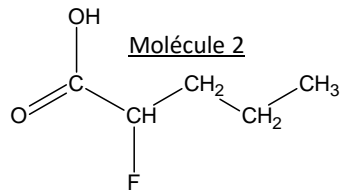
Molécule 3



2,3-diméthylbutan-2-ol

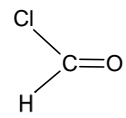
alcool tertiaire

Molécule 2



acide 2-fluoropentanoïque

Molécule 4



chlorométhanal