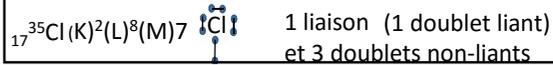
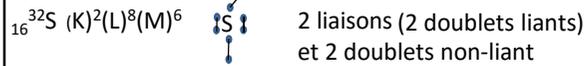
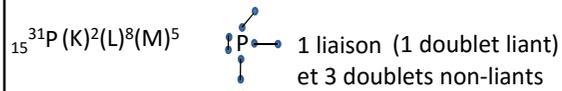
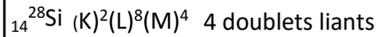
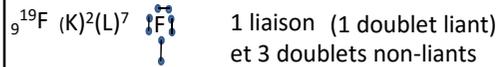
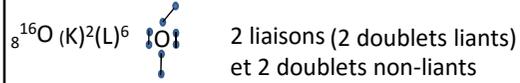


Correction exercices: Formule de Lewis avec doublets non-liants,

Exercice 1: **isomérisme Z/E et formule topologique.**

1. H: 1 doublet liant

C: 4 doublets liants



${}^5_{11}\text{B} (\text{K})^2(\text{L})^3$: cet atome ne peut pas respecter la règle de l'octet car ses 3 e⁻ vont former 3 liaisons en s'appariant avec 3 e⁻ «extérieur», pour conséquent, cet atome sera entouré de 6 e⁻ au lieu de 8.

